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TS WOC 2-2

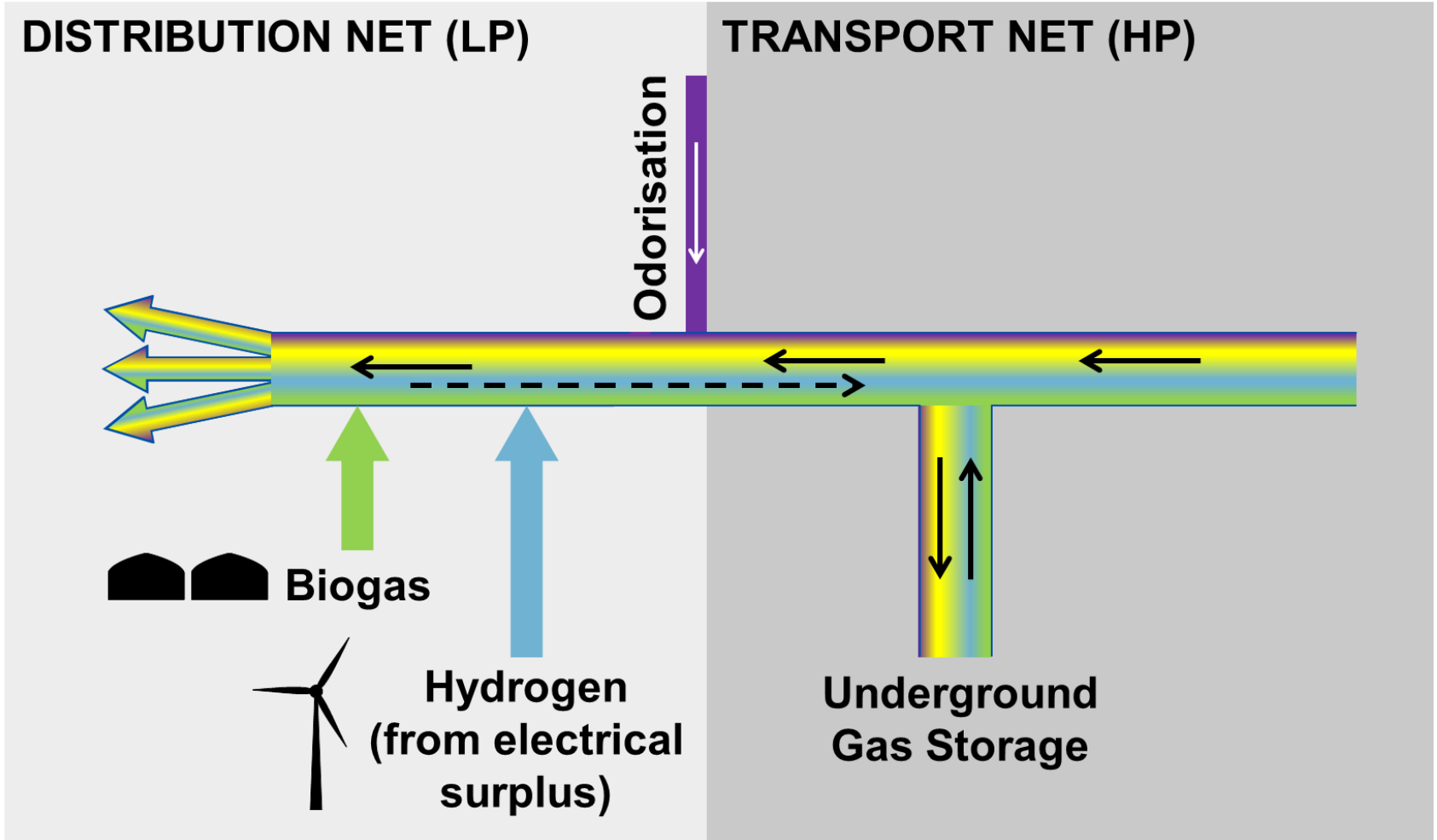
COMPATIBILITY OF NATURAL GAS
SUBSTITUTES FROM RENEWABLE ENERGY
SOURCES WITH UNDERGROUND GAS
STORAGES

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The old times are gone



Limit values for regenerative gases vs. natural gas

	Biogas	Hydrogen (H ₂)	German limit UGS	EASEE-Gas
O ₂ [mol%]	≤ 3 (30,000 ppm)	-	0.001 (10 ppm)	0.001 (10 ppm)
CO ₂ [mol%]	≤ 5	-	-	2.5
H ₂ [mol%]	-	≤ 100	-	-
Sulphur [mg/sm ³]	≤ 6 non odorised	≤ 6 non odorised	6	30
Odorisation	odorised	odorised	not odorised	not odorised
Microbiology	-	-	-	-

Effect renewables on underground gas storages

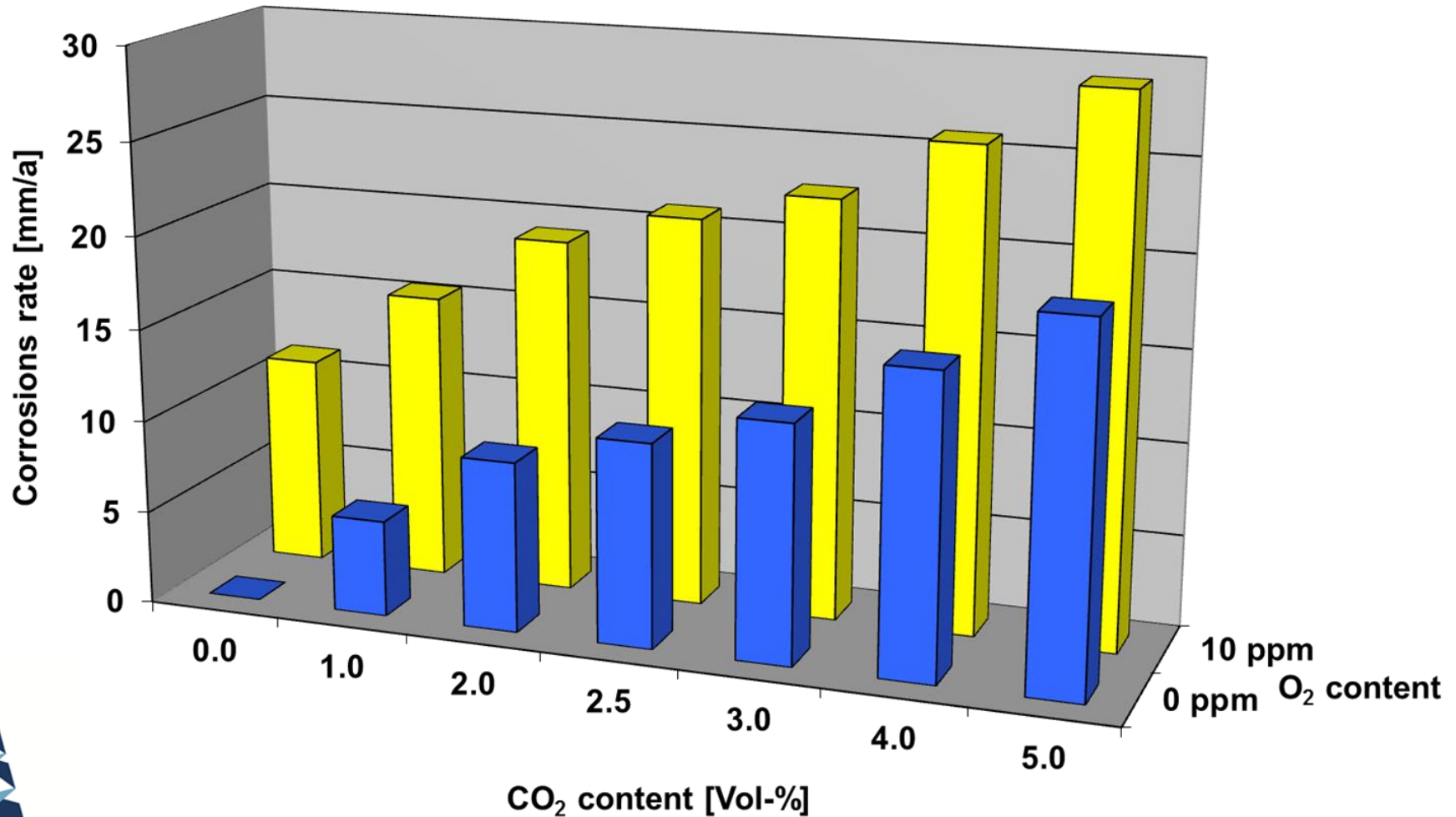
- Underground gas storages vital for secure energy supply in EU
- What is effect of renewable gases on
 - Reservoir rock ?
 - Subsurface installations?
 - Gas treatment ?
- → **5 literature studies to evaluate risks:**
 - Hydrogen (H₂) (DGMK project 752)
 - Oxygen (O₂) (DGMK project 753)
 - Carbon dioxide (CO₂) (DGMK project 754)
 - Sulphur and Odorant (DGMK project 755)
 - Microbiology (DGMK project 756)

Influence of carbon dioxide (CO₂)

- No significant influence on geology of UGS
- No negative impact on tightness of geological seal of UGS
- Potential H₂S liberation in reservoir by increased CO₂ content in injection gas
- No significant corrosion of cement expected
- Increases in steel corrosion with CO₂ content (next slide)
- **→ Keeping existing EASEE-Gas limit for CO₂ of 2.5 mol%**

Carbon dioxide and corrosion

Theo. corrosions rates for carbon steel $T = 45\text{ }^{\circ}\text{C}$, $p = 250\text{ bar}$

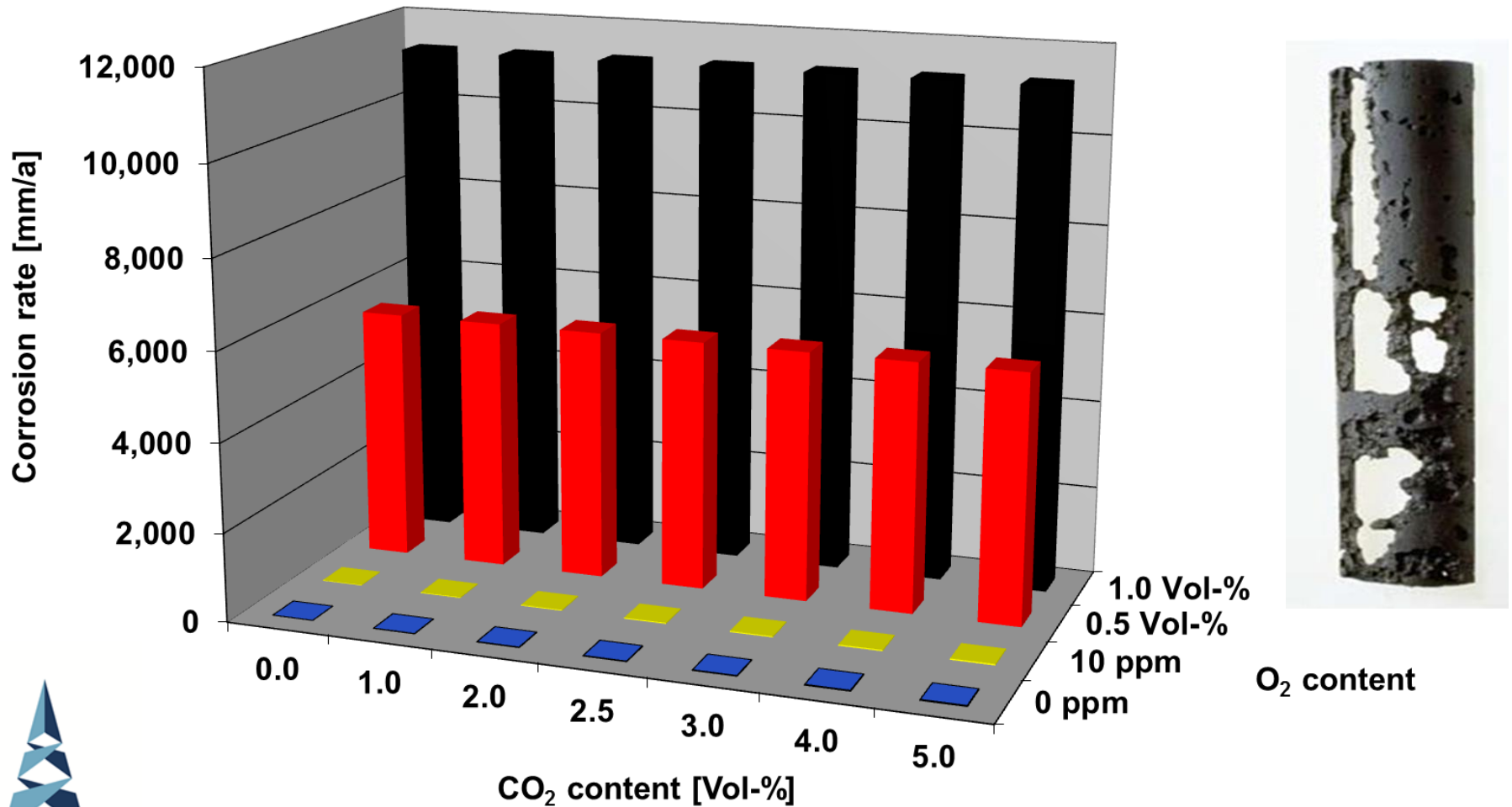


Sulphur compounds and odorants

- Solution of sulphur components & odorants in formation water
 - Problems in water treatment and disposal
 - Entry in gas treatment units
 - Glycol degeneration
- Odour trouble
 - Risk of need gas additional purification steps
- Potential H₂S liberation by well work overs
- → **No odorised gas in UGS**
- → **Keeping up existing limits for sulphur compounds**

Oxygen (O₂) and corrosion

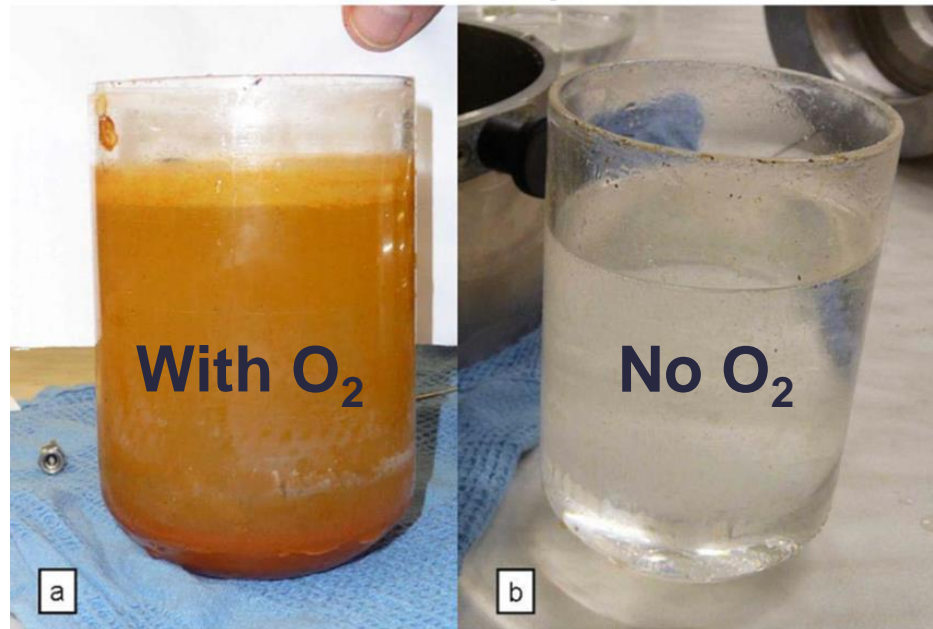
Theo. corrosions rates for carbon steel T = 45 °C, p = 250 bar



O₂ and formation damage

- Mineral transformation and precipitation, e.g. $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$

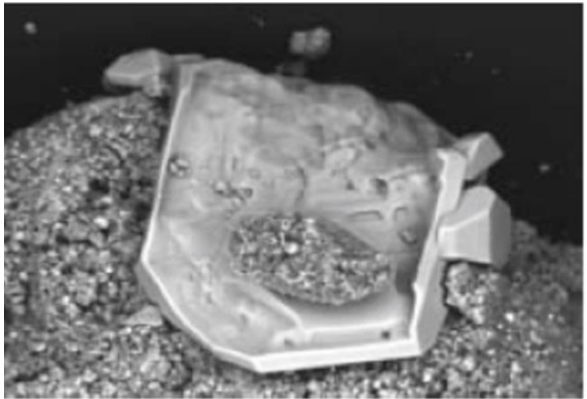
Fe²⁺ containing solution



- For 1 m³ Biogas (O₂: 2 %) blending with min. 2,000 m³ natural gas for reaching 10 ppm O₂ limit

O₂ and sulphur compounds

- Sulphur deposits by caused by O₂
 - Formation damage
 - Damage of installations
 - Localized corrosion



Sulphur crystal in pipeline generated by O₂



Sulphur deposits in subsurface valve



Pipeline corrosion by sulphur deposits

➔ Limitation of O₂ content to 10 ppm in injection gas

Microorganisms in biogas

- So far no pathological germs observed in biogas
- If sufficiently treated no other microorganisms (MO) expected in biogas than in natural gas
- Installation of micro filter as additional safety measure possible
 - Current practice in the Netherlands
- **→ Microbiology of biogas no concern if treated sufficiently**

Hydrogen (H₂) – the big unknown

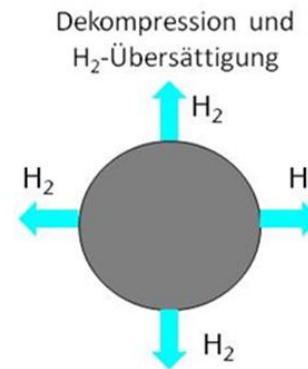
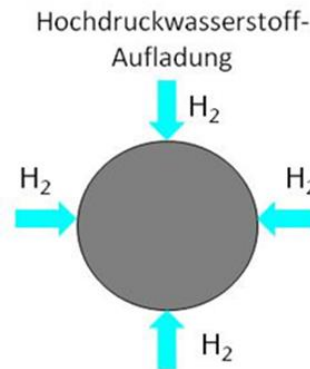
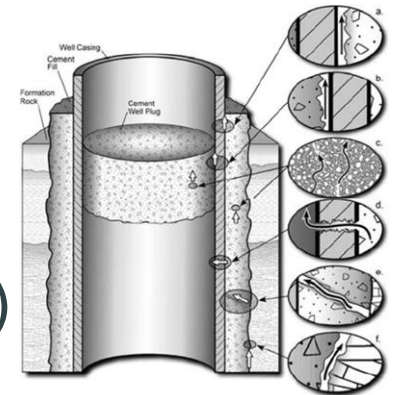
- Hydrogen (H₂) – different properties than methane
- No experiences with effect of H₂ on natural gas storages
- Currently no limit value for H₂ content in natural gas
- → **Many open questions**
 - → **Demand for research**
 - Geological integrity and amount of storable energy
 - Technical integrity
 - Effect on microbiology in UGS

H₂ – Integrity of storages & storable amount of energy

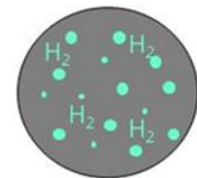
- H₂ content in gas up - amount of storable energy down
- Diffusion of H₂ through cap rock no risk
 - But losses by initial H₂ saturation of formation water
- Hydraulic integrity of the cap rock?
 - Modification surface tensions of stored gas and capillary bound water?
- Geochemical stability of reservoirs?
 - Reaction von H₂ with reservoir rock?
- → Porous rock storages more at risk than cavern storages

H₂ - Integrity of subsurface installations

- Influence on cementation
 - Modification recipes & additives necessary to prevent H₂ diffusion
- Influence on steel installations (e.g. tubings)
 - Critical: steels with strength ≥ 800 MPa
- Influence on elastomer based seals (e.g. packers)
 - Problematic: rapid pressure release



Innere Frakturen durch Freisetzung des Wasserstoffs



H₂ - Integrity of surface installations

- H₂ has other safety properties than natural gas
 - Lower ignition energy
 - Faster speed of flame
- H₂ ≤ 10Vol.-% so far no indications for critical safety issues
- Current H₂ tolerance of turbine compressors: 1 - 5% H₂
- Work of compression increases with increasing H₂ content

H₂ and microorganisms

- There are fully functioning H₂ – UGS, but
 - They have no carbon source needed for biological growth
 - Natural gas - hydrogen mixtures in contrast have a carbon source
- Experiences with underground storage of town gas
 - Town gas: mixture predominantly of CO, H₂ und CH₄
 - of CO, H₂ und CH₄
 - Significant microbiological activity observed

Change of gas quality by H₂ consumption

Tabelle 10: Gasqualitätsänderung im Horizontalmodell

1. Einlagerung: 7. 12. 84 - 8. 2. 85 63 Tage

Gasvolumenänderung: eingesetztes Stadtgasvolumen 400 ml

Gasverbrauch: 195 ml = 48,75 %

Restgas 205 ml

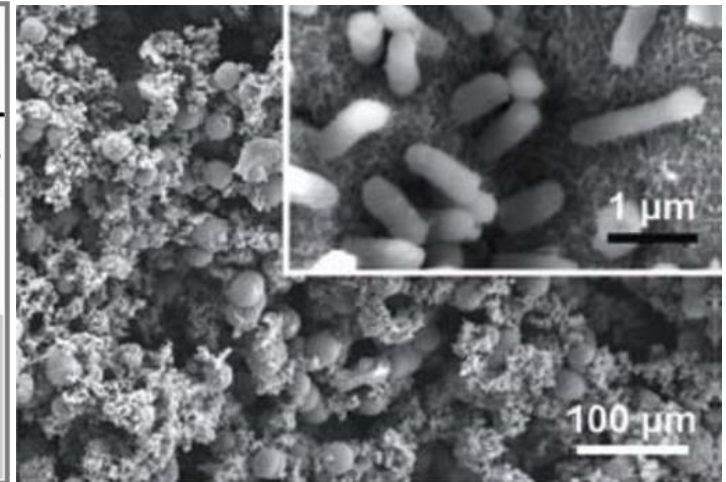
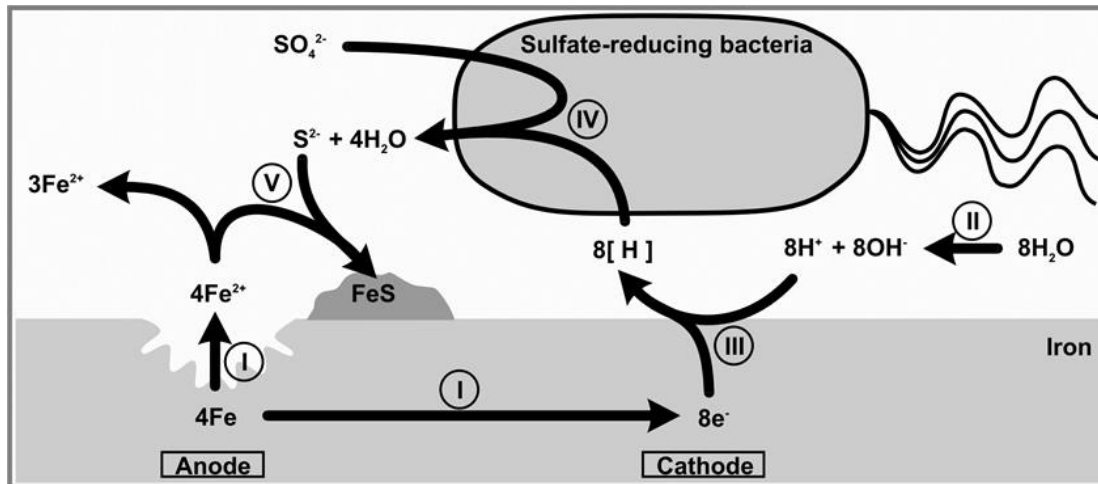
Loss of almost 50% of volume

Komponenten	Einspeisegas		Ausspeisegas		Gasvolumen- änderung	
	%	ml	%	ml	ml	%
CH ₄	29,5	118	34,2	70,11	-47,85	-40,55
C ₂ H ₆	0,37	1,48	0,36	0,738	- 0,742	-50,01
C ₃ H ₈	0,044	0,176	0,026	0,053	- 0,123	-69,89
i C ₄ H ₁₀	0,0041	0,016	0,0021	0,004	- 0,012	-75,0
n C ₄ H ₁₀	0,0055	0,022	0,0016	0,0033	- 0,019	-86,36
22 PM Pr	0,001	0,004	0,0005	0,001	- 0,003	-75
i C ₅ H ₁₂	0,0012	0,005	0,0001	0,0002	- 0,0048	-96
n C ₅ H ₁₂	0,001	0,004	10 ⁻⁴	0	- 0,004	-100
i C ₆ H ₁₄	0,0006	0,002	n.w.	0	- 0,002	-100
n C ₆ H ₁₄	0,0002	0,001	n.w.	0	- 0,001	-100
i N ₂	32,5	130	56,2	115,21	-14,79	-11,38
O ₂	0,31	1,24	0,2	0,41	- 0,83	-66,93
CO ₂	n.w.	0	8,4	17,22	+17,22	+
H ₂	24,2	96,8	0,05	0,102	-96,7	-99,9
He	0,12	0,48	n.w.	0	- 0,48	-
Ar	0,26	1,04	0,53	1,08	+ 0,04	- 3,85
CO	12,9	51,6	n.w.	0	-51,6	-100

Complete loss of H₂

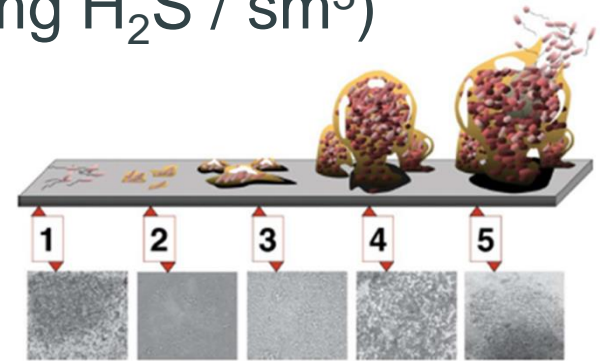
H₂ – Food for microorganisms

- $4\text{H}_2 + \text{SO}_4^{2-} + 2\text{H}^+ \rightarrow \text{H}_2\text{S} + 4\text{H}_2\text{O}$ - Sulfat Reducing Prokaryotes
- $2\text{CO}_2 + 4\text{H}_2 \rightarrow \text{CH}_3\text{COOH} + 2\text{H}_2\text{O}$ - Acetogenese
- $4\text{H}_2 + \text{CO}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O}$ - Methanogenese



H₂ – Effect of microbiological activity

- Corrosion
 - MIC (Microbiologically Influenced Corrosion)
 - Formation of H₂S und organic acids
- Risk of limit exceeded for natural gas (5 mg H₂S / sm³)
- Loss of permeability
 - Porous rock reservoirs only
- H₂ conversion in UGS
 - Loss of energy and change of gas composition
 - Loss of storage capacity by temperature increase



H₂ – Factors influencing microbiological activity

- Pressure – no great influence
- Temperature – optimum 30-50°C, but also higher temperatures suitable
- Salinity – The higher the worse for microbiological activity
- Surface area – The more the better for microbiological activity
 - Surface good basis for growth, adsorption and reaction
- **→ Porous rock storages more at risk than cavern storages**

Conclusions

- Underground gas storages vital part of gas infrastructure
- Change in gas composition by gases from renewable energies
- Damage of underground gas storages (UGS) must be avoided
- Necessary measures:
 - → **Biogas**
 - Keeping the existing limit values
 - Sufficient treatment of microbiology
 - → **Hydrogen (H₂)**
 - So far no limit values in place
 - Need for research to establish limit value
 - e.g. ‘Underground Sun Storage’ project

Co-authors

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